

Synthesis of  $\Delta^3$ -1,3,4-Telluradiazolines, a Novel Tellurium-Containing Heterocycle

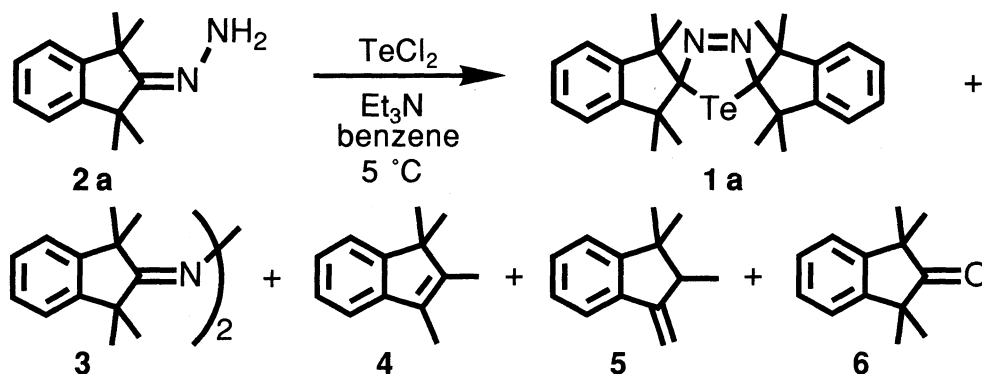
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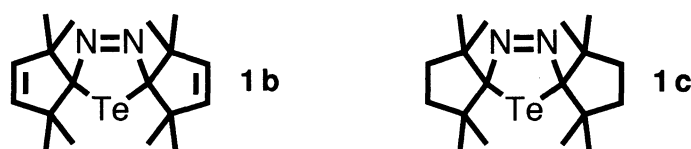
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Reaction of sterically hindered hydrazones with tellurium dichloride in the presence of triethylamine in benzene afforded  $\Delta^3$ -1,3,4-telluradiazolines, a novel heterocycle, probably via 1,3-dipolar cycloaddition of telluroketones with diazo compounds, both generated in situ.

The chemistry of  $\Delta^3$ -1,3,4-thia-<sup>1)</sup> and selenadiazolines,<sup>2)</sup> which can be synthesized by cycloaddition of diazo compounds with corresponding thio- and selenoketones respectively, have intensively been studied because of their usefulness in the preparation of intriguing molecules such as extremely sterically hindered olefins by thermal two-fold extrusion reactions.<sup>1,2)</sup> It is known that the selenadiazolines are more useful than the thiadiazolines in the two-fold extrusion methodology because of the greater magnitude of the atomic radius of selenium.<sup>2a)</sup> It is reasonably considered, therefore, that the tellurium analogue,  $\Delta^3$ -1,3,4-telluradiazoline, is more suitable, but the cycloaddition route to the telluradiazoline is impossible since no stable telluroketone has been known.<sup>3)</sup> We now describe here the first synthesis of a novel heterocycle,  $\Delta^3$ -1,3,4-telluradiazoline **1**, by a unique reaction of a hydrazone with tellurium dichloride.

Treatment of 1,1,3,3-tetramethylindanonone hydrazone (**2a**)<sup>2b)</sup> with tellurium dichloride<sup>4)</sup> in the presence of triethylamine gave telluradiazoline **1a** in 26% yield. A typical procedure for the preparation of **1a** follows. To a degassed benzene solution (20 ml) of triethylamine (243 mg, 2.40 mmol) and hydrazone **2a** (202 mg, 1.00 mmol) was added freshly prepared powdered tellurium dichloride (238 mg, 1.20 mmol) at 5 °C under an argon atmosphere in the dark, and the mixture was well stirred for 1 h. After the insoluble material was filtered off through a plug of Celite, the solvent was removed under reduced pressure. The residue was washed with pentane carefully to afford **1a** (65.2 mg, 26%).<sup>5)</sup> Chromatographic separation of pentane-soluble products (silica gel/hexane) gave azine **3** (8%), olefins **4** (32%) and **5** (16%), and ketone **6** (1%).

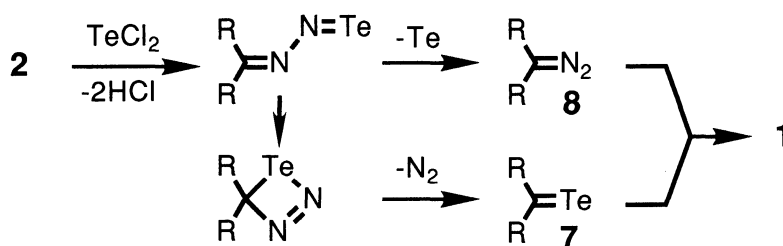




Telluradiazolines **1b** (10%) and **1c** (11%) were similarly synthesized from the corresponding hydrazones.<sup>5,6</sup> Telluradiazolines **1a-c** are thermally stable crystalline compounds though highly light sensitive even in the solid state. The spectral characteristic of **1** is considerably low chemical shifts of the spiro carbons in their <sup>13</sup>C-NMR spectra (133.2(**1a**), 128.7(**1b**), 130.0(**1c**)), which are shifted downfield compared with those of the corresponding thia-<sup>1c</sup>) and selenadiazolines.<sup>2b,c</sup>

Telluradiazolines **1** are most likely formed by 1,3-dipolar cycloaddition of telluroketone **7** with diazo compound **8**, both generated in situ, as shown in the following scheme. We previously reported similar reactions of hydrazones with sulfur chloride or selenium chloride resulting in the synthesis of stable thio- or selenoketones.<sup>7)</sup>

Further investigation on the reactivities of these novel telluradiazolines **1** is currently in progress.



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- 3) For the formation of transient telluroketones, see M. Segi, T. Koyama, Y. Takata, T. Nakajima, and S. Suga, *J. Am. Chem. Soc.*, **111**, 8749 (1989).
- 4) E. E. Aynsley, *J. Chem. Soc.*, **1953**, 3016.
- 5) **1a**: pale yellow needles, mp 162-164 °C (decomp); <sup>1</sup>H-NMR(CDCl<sub>3</sub>, 500 MHz) δ 1.13(s, 12H), 1.53(s, 12H), 7.21-7.26(m, 8H); <sup>13</sup>C-NMR(CDCl<sub>3</sub>, 125 MHz) δ 23.9(q), 37.4(q), 53.8(s), 122.8(d), 127.4(d), 133.2(s), 148.3(s); <sup>125</sup>Te-NMR(CDCl<sub>3</sub>, 85.1 MHz) δ -196.6; IR (KBr) 1591, 1577, 1481, 1450, 1376, 1364, 1313, 957, 841 cm<sup>-1</sup>; UV-vis(cyclohexane) λ<sub>max</sub> 340 nm (ε 640); HRMS (70 eV): Found: *m/z* 502.1641. Calcd for C<sub>26</sub>H<sub>32</sub>N<sub>2</sub><sup>130</sup>Te: 502.1628. **1b**: pale yellow needles, mp 115-117 °C (decomp); <sup>1</sup>H-NMR(CDCl<sub>3</sub>, 500 MHz) δ 0.83(s, 12H), 1.30(s, 12H), 5.79(s, 4H); <sup>13</sup>C-NMR(CDCl<sub>3</sub>, 125 MHz) δ 23.4(q), 34.9(q), 55.8(s), 128.7(s), 137.5(d); HRMS (70 eV): Found: *m/z* 402.1333. Calcd for C<sub>18</sub>H<sub>28</sub>N<sub>2</sub><sup>130</sup>Te: 402.1324. **1c**: pale yellow needles, mp 108-110 °C (decomp); <sup>1</sup>H-NMR(CDCl<sub>3</sub>, 500 MHz) δ 0.78(s, 12H), 1.23(s, 12H), 1.91-2.20(AA'BB', 8H); <sup>13</sup>C-NMR(CDCl<sub>3</sub>, 125 MHz) δ 26.8(q), 35.9(q), 39.1(t), 50.8(s), 130.0(s); HRMS (70 eV): Found: *m/z* 406.1625. Calcd for C<sub>18</sub>H<sub>32</sub>N<sub>2</sub><sup>130</sup>Te: 406.1627.
- 6) Compounds **1b** and **1c** were purified by column chromatography (silica gel/hexane). The lower yields for **1b** and **1c** than for **1a** are due to their decomposition during the chromatography.
- 7) R. Okazaki, K. Inoue, and N. Inamoto, *Tetrahedron Lett.*, **38**, 3673 (1979); R. Okazaki, K. Inoue, and N. Inamoto, *Bull. Chem. Soc. Jpn.*, **54**, 3541 (1981); R. Okazaki, A. Ishii, and N. Inamoto, *J. Chem. Soc., Chem. Commun.*, **1983**, 1429; A. Ishii, R. Okazaki, and N. Inamoto, *Bull. Chem. Soc. Jpn.*, **61**, 861 (1988).

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